Isolation and Structure of an Antimitotic Cyclic Peptide, Ustiloxin F: Chemical Interrelation with a Homologous Peptide, Ustiloxin B

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Ustiloxin F, a microtubule inhibitor, was isolated as a minor metabolite of *Ustilaginoidea virens*. The structure was determined from the spectral data and by chemical interrelation to ustiloxin B through reductive removal of the sulfoxide-containing side chain of ustiloxin B to give ustiloxin F. Ustiloxin F inhibited microtubule assembly with an IC_{50} value of $10.3 \, \mu M$.

Ustiloxins $A \sim D$ ($1 \sim 4$), which are strong inhibitors of microtubule assembly, have been isolated from the water extract of false smut balls on rice panicles caused by the fungus *Ustilaginoidea virens*, $^{1 \sim 3}$) and their structures were determined to be as shown in Figure 1.⁴) During our extensive search for unknown active principles in the water extract of false smut balls, ustiloxin F (5) was isolated as a new minor component. The structure of ustiloxin F was determined by spectroscopic analyses and by chemical interrelation with ustiloxin B (2).

This paper deals with isolation, structure determination and anti-tubulin activity of ustiloxin F (5), as well as the chemical conversions of ustiloxin A (1) to D (4) and ustiloxin B (2) to F (5) through a reductive cleavage of the sulfoxide-containing side chain on the aromatic ring.

Results and Discussion

The isolation procedure for ustiloxin F (5) is shown in Figure 2. The water extract of false smut balls was subjected to ODS column chromatography, and fractions containing ustiloxins B, D and F were combined. Ustiloxin F (5) was purified from the combined fraction by successive column chromatographies using Diaion CHP 20P, silica gel, ODS, and finally ODS-HPLC. The overall yield of 5 was less than 0.5 mg starting from 500 g

of false smut balls.

The molecular formula (C₂₁H₃₀N₄O₈) of 5 determined by FAB-MS and HRFAB-MS was C₅H₉NO₄S less than 2 and C₂H₄ less than 4 (Table 1). Its ¹H and ¹³C NMR data⁶⁾ suggested that compound 5 lacks the side chain on the aromatic ring, like ustiloxin D (4), and has the same cyclic peptide skeleton as ustiloxin B (2) (Tables 2 and 3). Combining these spectroscopic data, the structure of 5 was elucidated and was proved by direct conversion of ustiloxin B (2) to 5 via reductive removal of the sulfoxide-containing side chain of 2 by PtO₂ treatment in water.

In order to optimize the reaction conditions for cleavage of the sulfur-containing side chain, conversion of ustiloxin A (1), the most abundant homolog of this family, to ustiloxin D (4) was attempted under a variety of reduction conditions. The optimal result so far attained involved stirring a solution of 1 in H₂O for 3 hours in the presence of PtO₂ (15-fold molar excess) under atmospheric pressure of H₂ (Figure 3). The yield of 4 was 58 to 67% with 14% recovery of 1. In acetic acid with PtO₂, the yield of 4 after 5 hours was only 25 to 40% with the recovery of 30% of 1. In ethanol with PtO₂, the initially formed 4 was N-ethylated by acetal-dehyde generated *in situ* to give N-ethylustiloxin D.⁵⁾ In the case of the reaction with Raney Ni, 4 could not be isolated.

Based on the above result, hydrogenolysis of 2 was

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Fig. 1. Structures of ustiloxins.

$$HO_2C$$
 HO_2C
 HO_2

Ustiloxin A (1)^{1,2)}

Ustiloxin C (3)²⁾

Ustiloxin D (4)²⁾

Ustiloxin F (5)

carried out with PtO_2 (×15 molar) in H_2O for 4 hours to give 44% of 5 along with 22% recovery of 2. The synthetic 5 obtained from 2 was identical with natural 5 in its 1H NMR, FAB-MS and HPLC retention time. The signals in the NMR spectra of 4 and 5 were fully assigned with the aid of 1H , ^{13}C , COSY (4 only), C-H COSY and HMBC experiments (Tables 2 and 3). Isopropyl signals of 4 at positions 24, 25 and 26 appear at 1.80, 0.71,

0.80 ppm ($\delta_{\rm H}$) and 31.6, 21.3, 20.8 ppm ($\delta_{\rm C}$), respectively, while the methyl signal of 5 at position 24 appears at 1.05 ppm ($\delta_{\rm H}$) and 15.8 ppm ($\delta_{\rm C}$).

The inhibitory activity of 5 against tubulin polymerization was evaluated. The IC₅₀ value, *i.e.*, the concentration required for 50% inhibition of polymerization, was determined to be $10.3 \,\mu\text{M}$ (Figure 4).

The IC₅₀ values of ustiloxin A (1), B (2), D (4) and

Fig. 2. Isolation of ustiloxin F from water extract of false smut balls.

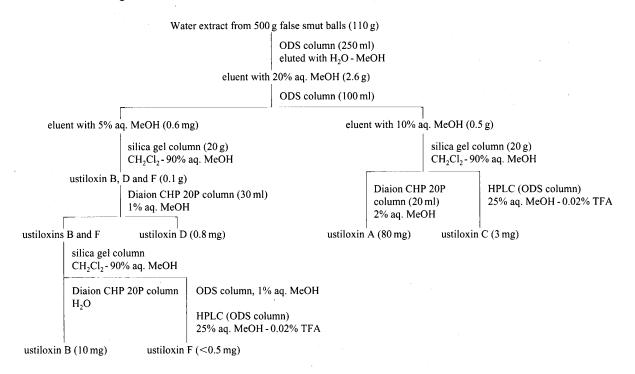


Table 1. Physico-chemical properties of ustiloxins B (2), D (4) and F (5).

	Ustiloxin B ²⁾	Ustiloxin D	Ustiloxin F
Molecular formula	$C_{26}H_{39}N_5O_{12}S$	C ₂₃ H ₃₄ N ₄ O ₈	$C_{21}H_{30}N_4O_8$
Appearance	Colorless powder	Colorless powder	Colorless powder
$UV \lambda_{max}^{H_2O} nm (\varepsilon)$	252 (5,000), 290 (2,500)	227 (6,200), 280 (1,600)	227 (4,500), 280 (1,100)
FAB-MS (m/z) $(M+H)$	646	495	467
HRFAB-MS (m/z) $(M+H)$			
Calcd	646.2394	495.2455	467.2142
Found	646.2390	495.2401	467.2186
[α] _D	$+14.1^{\circ}$ (c 0.5, H ₂ O)	-48.8° (c 0.5, H ₂ O)	-68.0° (c 0.5, H ₂ O)

Fig. 3. Hydrogenolysis of ustiloxin A (1) and ustiloxin B (2).

$$HO_2C$$
 HO_2C
 HO_2

Ustiloxin A (1) R=CH(CH₃)₂ Ustiloxin B (2) R=CH₃ Ustiloxin D (4) R=CH(CH₃)₂ Ustiloxin F (5) R=CH₃

Table 2. ¹H NMR chemical shifts (δ_H , ppm relative to water: δ_H 4.65 ppm), multiplicities, and coupling constants (J, Hz) of ustiloxins B (2), D (4) and F (5) and FGHMBC correlation of 4 and 5 in D₂O.

II NI.	2 ²⁾	4		5	
H No.	¹H NMR	¹H NMR	FGHMBC (C No.) ¹ H NMR FGHMBC	FGHMBC (C No.)	
3	4.55 s	4.81 s	2, 17	4.58 s	2, 5, 17, 21
6	4.27 q, 7.0	3.96 d, 10.0	5, 8, 24, 25, 26	4.24 q, 7.0	5, 8, 24
9	4.00 d, 10.0	3.34 d, 9.0	8, 10, N-CH ₃	3.26 d, 9.0	8, 10, 11, N-CH ₃
10	4.70 d, 10.0	4.50 d, 9.0	9, 11, 12, 16	4.51 d, 9.0	8, 9, 11, 12, 16
12		7.09 dd, 9.0, 2.0	10, 14, 16	7.04 dd, 8.3, 2.0	10, 14, 16
13	7.37 s	6.98 d, 9.0	11, 14, 15	6.89 d, 8.3	11, 14, 15
16	7.18 s	6.89 d, 2.0	10, 12, 14, 15	7.07 br s	12, 14, 15
19	3.60 d, 17.0	3.61 s	17, 20	3.66 d, 17.0	17, 20
	3.76 d, 17.0		,	3.71 d, 17.0	17, 20
21	1.58 s	1.63 s	2, 3, 22	1.54 s	2, 3, 22
22a	1.53 dq, 14.0, 7.2	1.65 dq, 15.0, 7.5	23	1.58 m ^b	2, 3, 21, 23
22b	1.95 dq, 14.0, 7.2	2.08 dq, 15.0, 7.5	2, 3, 23	1.91 dq, 14.0, 7.0	2, 3, 21, 23
23	0.82 t, 7.2	1.04 t, 7.5	2, 22	0.85 t 7.0	2, 22
24	1.04 d, 7.0	1.80 m	,	1.05 d, 7.0	5, 6
25		0.71 d, 6.5	6, 24, 26		
26		0.80 d, 6.5	6, 24, 25		
N-CH ₃	2.50 s	2.34 s	9	2.28 s	9
2'a	2.86 dd, 13.4, 2.8				
2′b	3.20 dd, 13.4, 10.0			_	
3′	4.19 dddd, 2.8, 3.0,	_		_	
	10.0, 10.0				
4'a	1.91 ddd, 3.0, 8.0, 15.0			_	
4′b	2.02 ddd, 4.0, 10.0, 15.0				
5′	3.80 dd, 4.0, 8.0				

Multiplicities, s: singlet, d: doublet, t: triplet, q: quartet, m: multiplet.

F (5) were compared under identical experimental conditions, and were obtained as 1.0, 1.8, 2.5 and $10.3 \,\mu\text{M}$, respectively. These values are somewhat different from those reported previously²⁾, and this may be partly due to inevitable slight differences in the experimental conditions and also in the purities of the samples used.

These values indicate that ustiloxin F, the smallest homolog of ustiloxins, retains the essential activity, but the functional groups of the side chains and the alkyl moiety on the cyclic skeleton enhance the binding affinity to tubulin.

Experimental

General

¹H and ¹³C NMR spectra were measured in H₂O on a JEOL ALPHA-500 NMR spectrometer at 500 and 125 MHz, respectively. FAB-MS, HRFAB-MS and CAD spectra were measured on a JEOL JMS-HX110 instrument with the EBE arrangement. Optical rotations were measured on a JASCO DIP-1000 digital polarimeter. UV

spectra were measured on a Shimadzu apparatus, model UV-2500. Thin layer chromatography was carried out on Merck Kieselgel 60F-254 plates, and HPLC was performed on a Shimadzu LC-10A apparatus.

Preparation of Microtubule Protein and Microtubule Assembly Assay

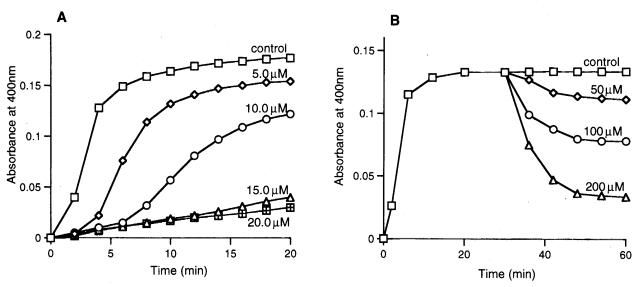
The procedures were the same as described in the previous papers.^{2,3)}

Conversion of Ustiloxin A (1) to Ustiloxin D (4)

A mixture of $100 \,\mathrm{mg}$ of compound 1 and PtO_2 (506 mg, $\times 15 \,\mathrm{molar}$) in water (10 ml) was stirred for 3 hours under an atmosphere of H_2 . The catalyst dissolved during the reaction. After removal of the water from the black solution with a rotary evaporator at $ca.50^{\circ}\mathrm{C}$, the products were extracted with 20 ml of acetic acid, with the aid of ultrasonication. The catalyst in the solution was separated by centrifugation, and the acetic acid was removed with a rotary evaporator. Compound 4 was purified by CHP 20P column chromatography (elution

b Overlapped with other signals.

Fig. 4. Effect of ustiloxin F on the polymerization and depolymerization of microtubule proteins.



- (A) Various concentrations of ustiloxin F were mixed with microtubule proteins (1.5 mg/ml) at 0° C and incubated at 37° C.
 - (B) Microtubule proteins (1.5 mg/ml) were incubated at 37°C.

Various concentrations of ustiloxin F were added 30 minutes later.

Table 3. ¹³C NMR chemical shifts (δ_C , ppm relative to dioxane δ_C 67.5 ppm) of ustiloxins B (2), D (4) and F (5) in D₂O.

C No.	2 ²⁾	4 ⁶⁾	5
2	84.5	88.7	86.2
3	60.3	62.3	60.3
5	172.7	173.8	173.1
6	50.0	62.8	49.7
8	169.4	168.9	171.9
9	68.2	71.7	71.1
10	74.7	75.9	74.4
11	129.0	133.3	132.5
12	137.1	125.3	123.2
13	114.4	121.7	118.8
14	153.2	153.6	150.7
15	146.3	145.5	142.7
16	124.6	126.4	124.0
17	170.6	173.5	170.9
19	44.2	46.6	44.2
20	176.9	179.0	176.9
21	22.2	24.0	22.0
22	31.7	35.2	31.6
23	8.4	10.7	8.4
24	15.7	31.6	15.8
25		21.3	
26		20.8	
N-CH ₃	33.0	35.2	33.6
2'	65.1		
3′	64.2		
4′	37.1		_
· 5′	53.2	•	
6'	174.8	 ,	

with 20% aqueous MeOH). 4: $40 \sim 47 \text{ mg}$ ($54 \sim 64\%$). Recovery of 1: 14%.

Conversion of Ustiloxin B (2) to Ustiloxin F (5)

Compound 2 was transformed to 5 by the same procedure as used in the synthesis of 4 from 1. The reaction rate of 2 to 5 was slower, and after 4 hours of reaction, compound 5 was obtained in 44% yield (32 mg) and the recovery of 2 was 22%. Compound 5 was purified by CHP 20P column chromatography (elution with water).

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- 6) ¹³C NMR data previously reported for ustiloxin D (4)²⁾ were revised.